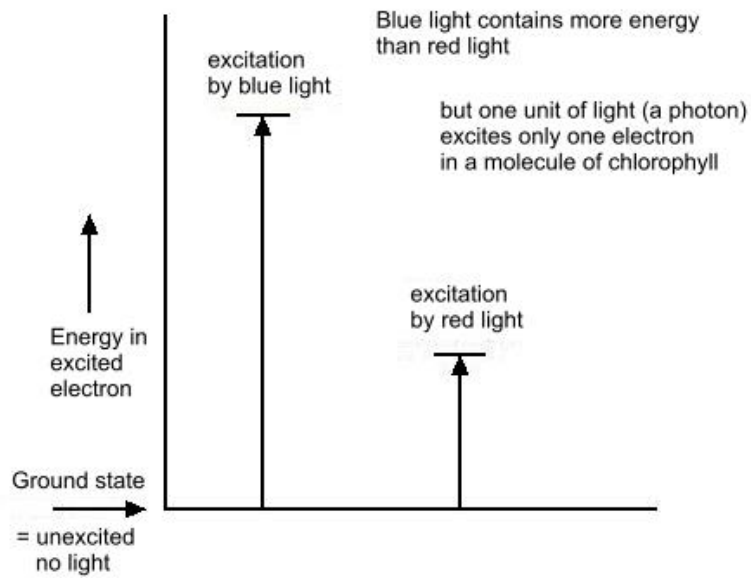


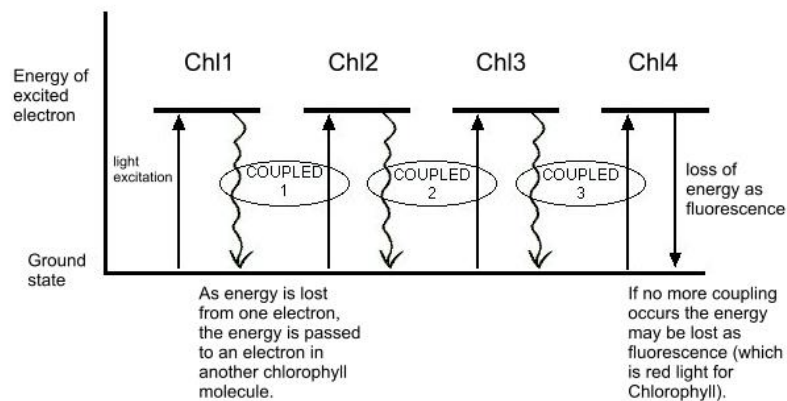
Light absorbed by chlorophyll excites the electrons in the ring as shown above. Different wavelengths of light excite the electrons by different amounts (Figure 2).

Energy Diagram



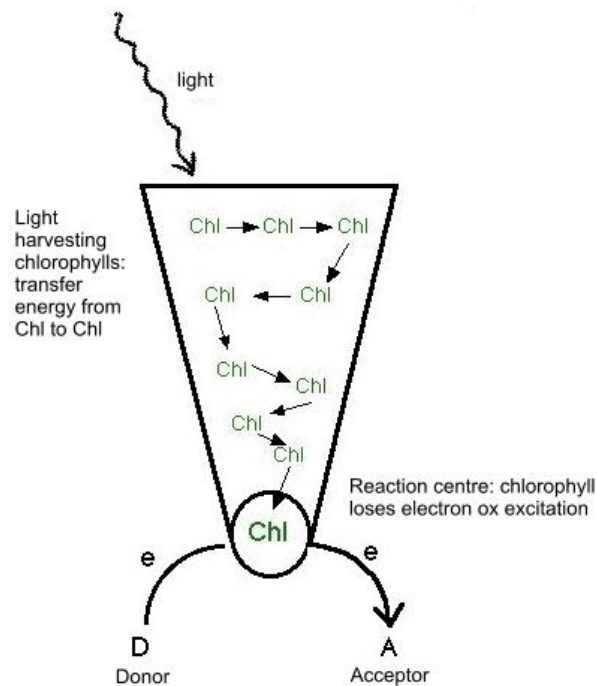
The energy in the 'excited electrons' can be passed from one chlorophyll molecule to another, but in the end it will just be lost as fluorescence (Figure 3) (ie the energy will be re-emitted as light), unless the excited electron itself can be ejected from the chlorophyll molecule.

Energy diagram for lots of chlorophyll molecules



This process of electron ejection takes place only in chlorophyll molecules which are specifically held in a special protein complex called a reaction centre (Figure 4).

Reaction centres - where electrons are ejected



Lost electron must be replaced from an electron donor

There are two different sorts of reaction centres in plants (Figure 5). In each of these reaction centres, the ejected electron is transferred to an acceptor molecule, which can then pass it on to a different molecule and eventually the electron(s) can be used to fix carbon dioxide. However, you can't keep on ejecting electrons from these special chlorophyll molecules, electrons must be fed back in to replace those ejected. These electrons come from water, resulting in oxygen being evolved.

Photosynthesis - 2 different Reaction Centres

